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MCQ Questions for Class 11 Chemistry Chapter 2 Structure ...

Electrons in the atom do not enter the nucleus. In fact, electrons in the s states tend to peak at the nucleus. Electrons are not little balls that can fall into the nucleus under electrostatic attraction. Rather, electrons are quantized wavefunctions that spread out in space and can sometimes act like particles in limited ways.

Why don't electrons in the atom enter the nucleus ...

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