

## File Type PDF Behavior Of Gases Webquest Answer Key

# Behavior Of Gases Webquest Answer Key

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## **Behavior Of Gases Webquest Answer**

The Behavior of Gases WebQuest Name \_\_\_\_\_ Period \_\_\_\_\_ Date \_\_\_\_\_  
1. What is a gas? 2. What does the kinetic molecular theory explain? 3. Describe the three states of matter. 4. Describe the typical motions of atoms in a solid, a liquid and a gas. 5. The Kelvin scale is based on the concept of absolute zero.

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Behavior of Gases Webquests. DIRECTIONS On this quest, you will be working in teams of 2, ... Data Recorder - records all data NEATLY in your packet (teammate will copy answers later). At the end of this WebQuest, there will be 2 evaluations. The first evaluation is a team evaluation.

## **Behavior of Gases Webquests - LAURA'S CLASSROOM**

Answer the following questions: a. What is the name of the gas law investigated? b. What is the value of the constant (R)? c. Which variables are inversely related? d. Which variable are directly proportional? e. What is the formula of the law? 5. Complete "Behavior of Gases" worksheet #8-10 . Title: Gas Laws Web-Quest

## **Gas Laws Web-Quest**

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provided to find the answers to these questions about the behavior of gas. [Filename: Student GasLawsWebQuest.pdf] - Read File Online - Report Abuse

## **Behavior Of Gases Web Quest Answers - Free PDF File Sharing**

That's called an "ideal gas". The behavior of real gases is quite similar to an ideal gas, except when the pressure is too high, or the temperature too low. That's called an "ideal gas".

## **Behavior of gases? - Answers**

some Characteristics of Gases, Liquids and solids and the Microscopic Explanation for the Behavior gas liquid \_ ooe 28. Use the chart to identify the state of matter described by the following. Many of these have more than one answer! (Use S, L or G in the spaces.) L not easily compressible rigid — particles locked into place flows easily

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## **The Behavior Of Gases Webquest**

Behavior Of Gases Webquest Answer Key Gas Laws Worksheet  
 $\text{atm} = 760.0 \text{ mm Hg} = 101.3 \text{ kPa} = 760.0 \text{ torr}$  Boyle's Law  
Problems: 1. If 22.5 L of nitrogen at 748 mm Hg are compressed to 725 mm Hg at constant temperature.

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