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Stoichiometry problems of this type are called either mass-volume or volume-mass problems. Because both types of problems involve a conversion from either moles of gas to volume or vice-versa, we can use the molar volume of 22.4 L/mol provided that the conditions for the reaction are STP.

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Simple stoichiometry only (one given, one wanted) Limiting reagents only (two given reactants, one wanted product) Mix & match (both simple stoichiometry and limiting reagent problems) Units to use (select at least one): Grams Moles Particles (e.g. atoms/molecules/formula units) Chemical formulas or names: Formulas only Names only

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